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Stereoselective Synthesis of (E,E)-1-Arylselenobutadienes by Cross-Coupling Reactions in the Presence of Palladium Catalyst

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Abstract: Hydrozirconation of arylselenoethynes 1 gives selenium-containing zirconium (IV) complexes 2, which are cross-coupled with alkenyl halides in the presence of Pd(PPh₃)₄ to afford (E,E)-1-arylselenobutadienes 4 in high yields.

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The stereocontrolled synthesis of conjugated dienes is of considerable interest in organic synthesis since such dienes are often encountered in natural compounds, such as Achillea amide¹, and are also valuable intermediates in the synthesis of more complex targets via Diels-Alder reaction².

In literature, we found the stereoselective synthesis of (E, E)-1-trimethylsi-lylbutadienes^{3,4}, (E, E)-1-phenylthiobutadienes⁵. However, there are few reports on the synthesis of (E, E)-1-arylselenobutadienes⁶.

Hydrozirconation⁷ can tolerate certain ether functionalities, such as OEt and OTHP group⁸, and the (E)-1-alkenylzirconium compounds thus obtained can react with alkenyl halides in the presence of catalytic amount of Pd(PPh₃)₄ or Ni(PPh₃)₄ to form 1,3-butadienes containing such functional groups^{8,9}.

We tried reacting arylselenoethynes 1^{10} with $Cp_2Zr(H)Cl$ and found that selenium-containing zirconium (W) complexes 2 were formed. Cross-coupling reaction of complexes 2 with alkenyl halides 3 in the presence of $Pd(PPh_3)_4$ afforded (E,E)-1-arylselenobutadienes 4 stereoselectively in high yields. We found that the optimum molar ratio of $Cp_2Zr(H)Cl$ to compound 1 was 1:1. The experimental results are summarized in Table 1.

Scheme 1

Ar in 1	3		Product 4	Yield(%)*
	R	X	_	
Ph	Ph	Br	4a	82
Ph	Ph	I	4a	77
Ph	CH ₃ OCH ₂	Br	4b	86
4-MeC_6H_4	Ph	Br	4c	80
$4\text{-MeC}_6\text{H}_4$	Ph	I	4 c	79
$4-MeC_6H_4$	CH₃OCH₂	Br	4d	88

Table 1 Synthesis of (E,E)-1-Arylselenobutadienes 4a-d

The alkenyl halides were prepared by hydrozirconation of 1-alkynes followed by trapping with electrophilic halogenation (I₂ or N-bromosuccinimide)¹³.

The major advantages of this coupling reaction are the preparation convenience of (E)-2-arylselenoalkenyl zirconium complexes 2 via hydrozirconation of arylselenoethynes 1 and the configuration retention of both the starting alkenyl zirconium complexes and the alkenyl halides.

Vinylic selenides can couple with Grignard reagents in the presence of catalytic amount of nickel-phosphine complexes to afford the corresponding unsaturated hydroarbons with loss of selenium-containing groups¹⁴. We reacted (E, E)-1-arylselenobutadienes 4 with Grignard reagents in the presence of NiCl₂(PPh₃)₂ and found that (E, E)-1, 4-disubstituted-1,3-butadienes 5 were formed, which are shown in scheme 3. The investigation on the synthetic applications of (E, E)-1-arylselenobutadienes 4 is in progress.

ArSe H
$$+ Ar^1MgBr \xrightarrow{THF} Ar^1 H H R$$
 H $+ Ar^1MgBr \xrightarrow{Cat, NICl_2(PPh_3)_2} H = H$

Scheme 2

a. isolated yield b. The configuration of double bonds were determined by ¹HNMR^{11,12}.

Table 2 Synthesis of (E,E)-1,4-disubstituted-1,3-butadienes Sa and Sb						
Product	R	\mathbf{Ar}^{1}	m.p.(°C) ^{a,b}	Yield(%)°,d		
5a	Ph	Ph	147-148 (149-150)	82		
5b	Ph	4-MeC ₆ H ₄	154-155 (155-156)	77		
literature ¹⁵ ; c.	isolated yield	ting points in parent ; d. NiCl ₂ (PPh ₃ nd arylmagnesium br	$)_2(3\%\times 0.5 \text{mmol})$), (E,E)-1-arylse-		

Synthesis of (F F)-1. A-disubstituted-1. 3-butadienes 59 and 5b

at r. t. for 48h to afford (E,E)-1,4-disubstituted-1,3-butadienes 5.

EXPERIMENTAL

¹H NMR spectra were recorded on a Bruker AC-P200(200M Hz) Spectrometer with Me₄Si as the internal standard in CDCl₃. Mass spectra were obtained on a HP 5890 A mass spectrometer and IR spectra on a Shimadzu IR-435 instrument. Microanalyses were measured using a Yanaco MT-3 CHN microelemental analyser. Solvent THF was distilled from sodium-benzophenone ketyl before use. Cp₂Zr(H)Cl was prepared according to literature¹³. All reactions were carried out under nitrogen.

General procedure for the synthesis of 4a-d:

To a dry 10ml flask charged with Cp₂Zr(H)Cl (0.8mmol) was injected THF (3ml), followed by the addition of 1 (0.8mmol) at 0°C. The mixture was stirred at 0°C for 40 min to yield a clear solution. It was then added alkenyl halide 3 (0.6mmol) and Pd(PPh₃)₄(0.6 ×5% mmol) and stirred at room temperature for 3 hrs. The resuting mixture was diluted with diethyl ether and after 5 min of additional stirring, the supernatant was filtered through a short plug of silica gel. After removal of solvent, the residue was purified by preparative TLC on silica gel (petroleum ether as eluent for 4a and 4c, diethyl ether/ petroleum ether (1:20) for 4b and 4d.

(E,E)-4-Phenyl-1-phenylseleno-1, 3-butadiene (4a):

m.p. $40-41^{\circ}$; IR(KBr) $v(cm^{-1})$; $3040,1593,981,736,690,^{1}$ H NMR $\delta(ppm)$; 724-735(m, 10H), 6. 40-7. 12(m, 4H); MS m/z; 286 $(M^+, 21)$, 128(100); Anal. Calcd. for $C_{16}H_{14}Se$: C, 67. 38; H, 4. 95. Found C, 67. 45; H, 5. 10.

(E,E)-5-Methoxyl-1-phenylseleno-1, 3-pentadiene (4b):

 $IR(film) \ \upsilon(cm^{-1}): 3032,2904,1570,734,686; \ ^{1}H \ NMR \ \delta(ppm): 7.25-7.51(m,5H),$ 6. 24-6. 73(m,3H), 5. 64-5. 71(m,1H), 3. 94(d,2H), 3. 31(s,3H); MS m/z: 254 $(M^+,11)$, 97(100); Anal. Calcd. for $C_{12}H_{14}OSe$: C,56. 94; H,5. 57. Found C,56. 83; H,5.72.

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(E,E)-4-Phenyl-1-(4-methyl phenylseleno)-1, 3-butadiene (4c):

IR (film) υ (cm⁻¹): 3015,1594,970,803,687; ¹H NMR δ (ppm): 7.06-7.49 (m,9H), 6.45-6.86 (m,4H), 2.39(s,3H); MS m/z: 300 (M⁺,26),128 (100); Anal. Calcd. for $C_{17}H_{16}Se$: C,68.24; H,5.39. Found C,68.31; H,5.54.

(E,E)-5-Methoxyl-1-(4-methylphenylseleno)-1, 3-pentadiene (4d):

IR(film) υ (cm⁻¹): 3012,2908,1600,980,804; ¹H NMR δ (ppm): 7. 12-7. 45(q,4H), 6. 20-6.71(m,3H), 5. 63-5.71(m,1H), 3. 95(d,2H), 3. 35(s,3H), 2. 27(s,3H); MS m/z: 268(M⁺,20),97(100); Anal. Calcd. for C₁₃H₁₆OSe: C,58. 44; H,6. 04; Found C, 58. 67; H,6. 11.

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